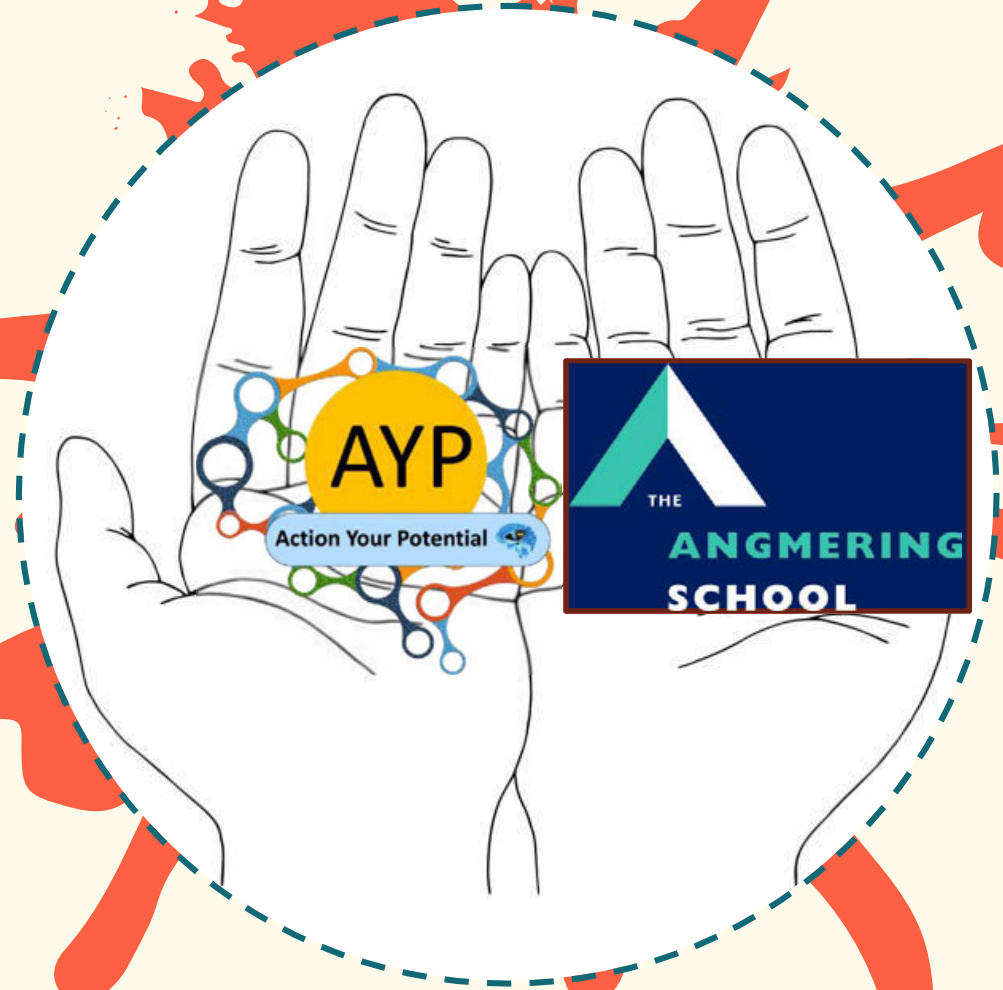
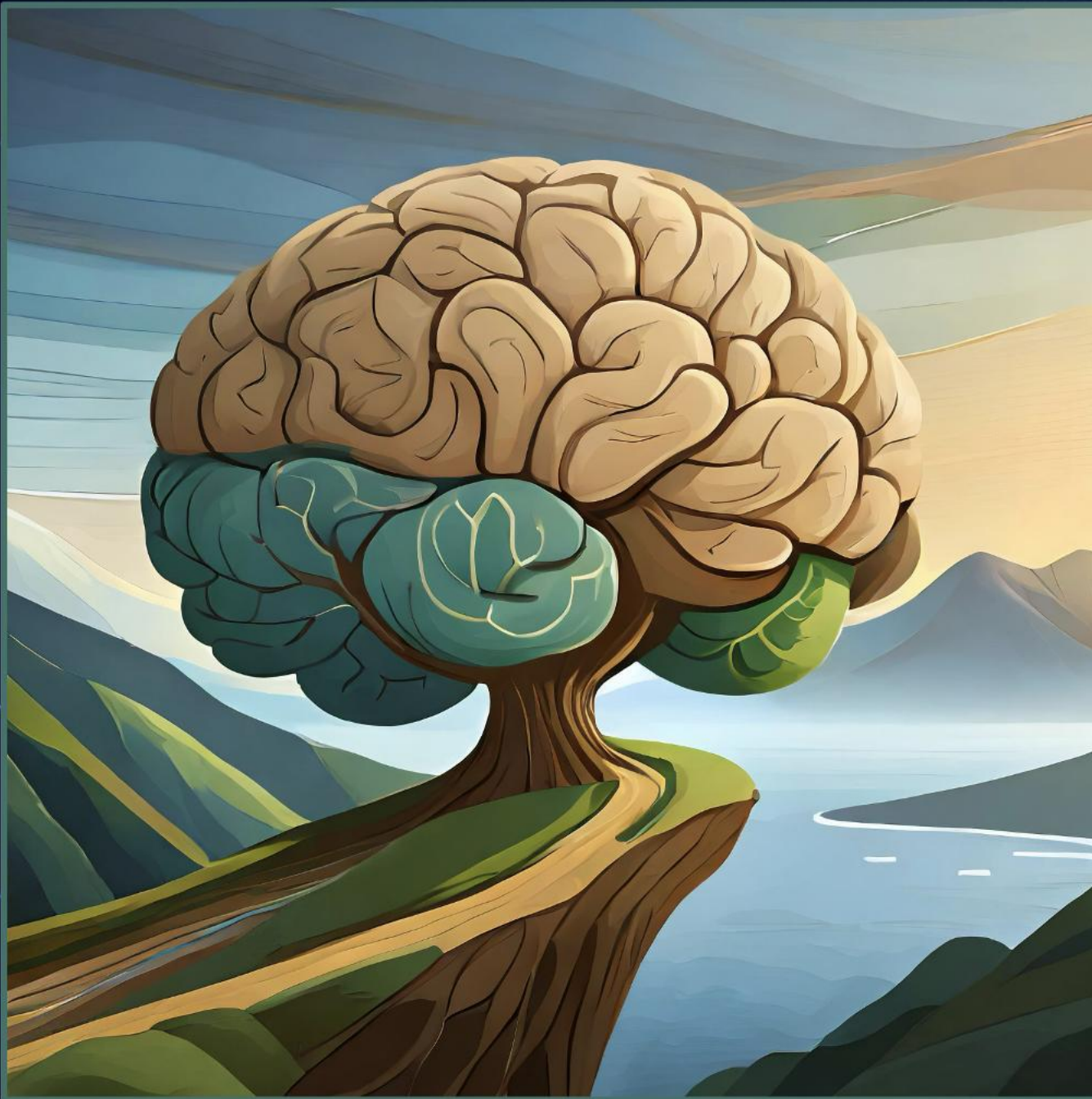


# AYP Support For Angmering School Students, Staff, Parents & Carers...



Our mission as an  
AYP Team...

...saving the world one brain at a time





$$1/5 + 3/2 = 4/6$$

$$\sqrt{85}$$



We're here to solve  
the GCSE problem

Become a #NeuroNinja  
this year and change  
your world



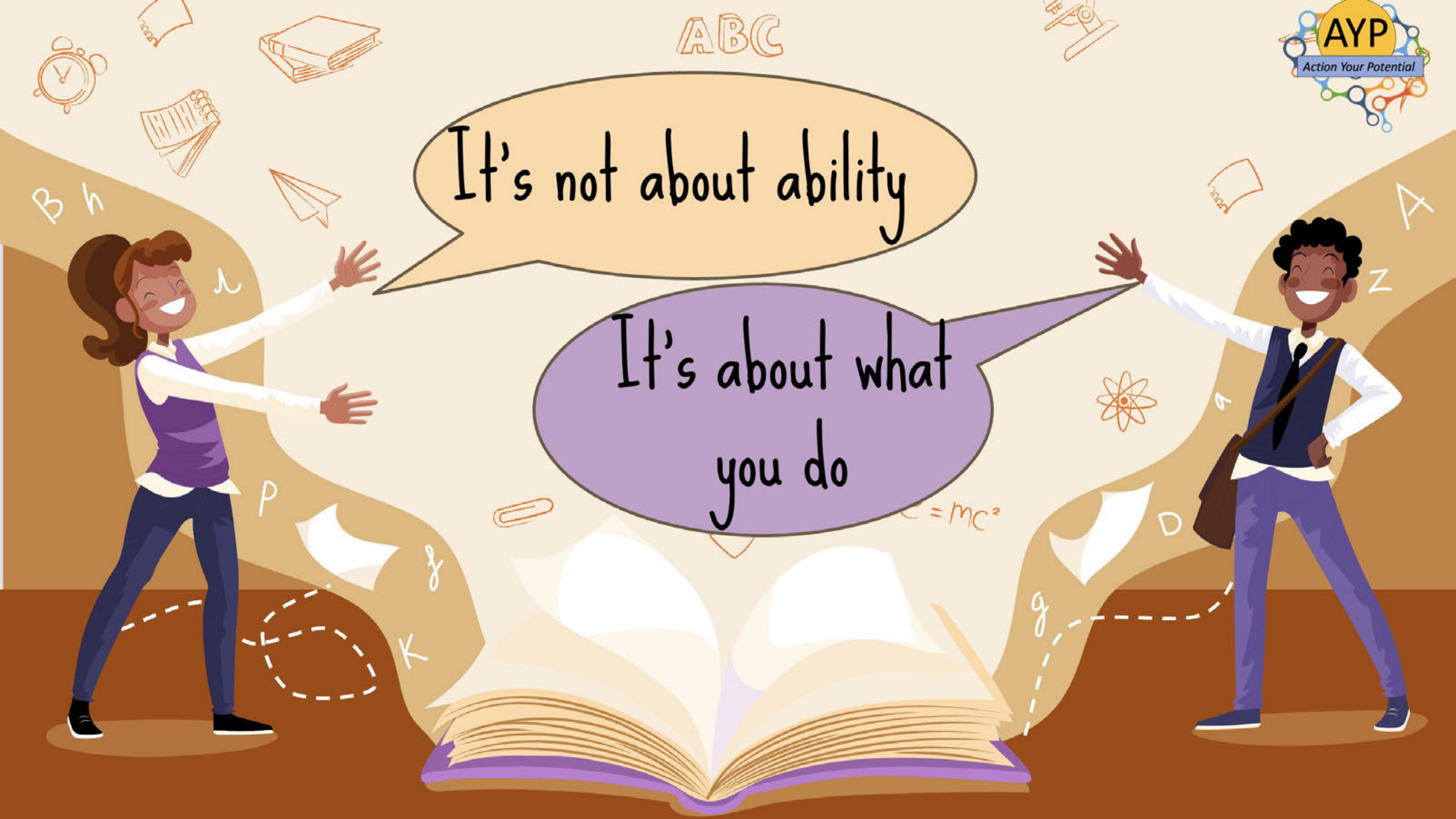
$$a^2 + b^2 = \sqrt{3}c \rightarrow AB + 7$$

ABC

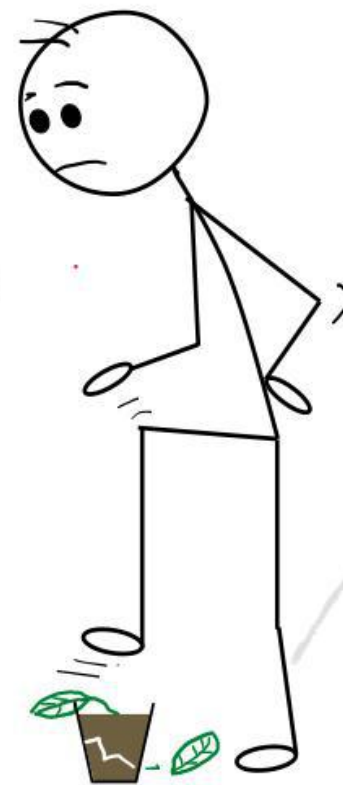
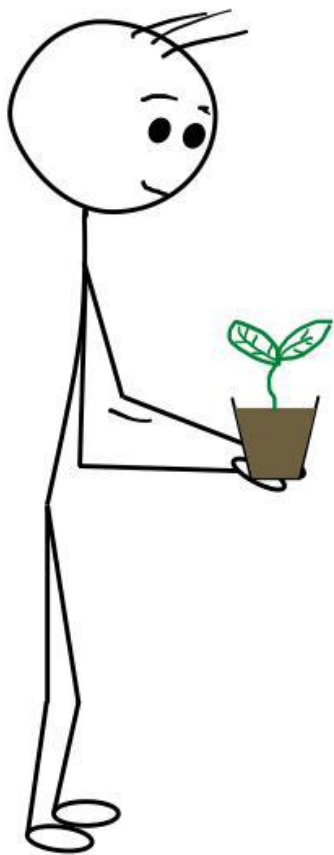
It's not about ability

It's about what  
you do

$$E=mc^2$$

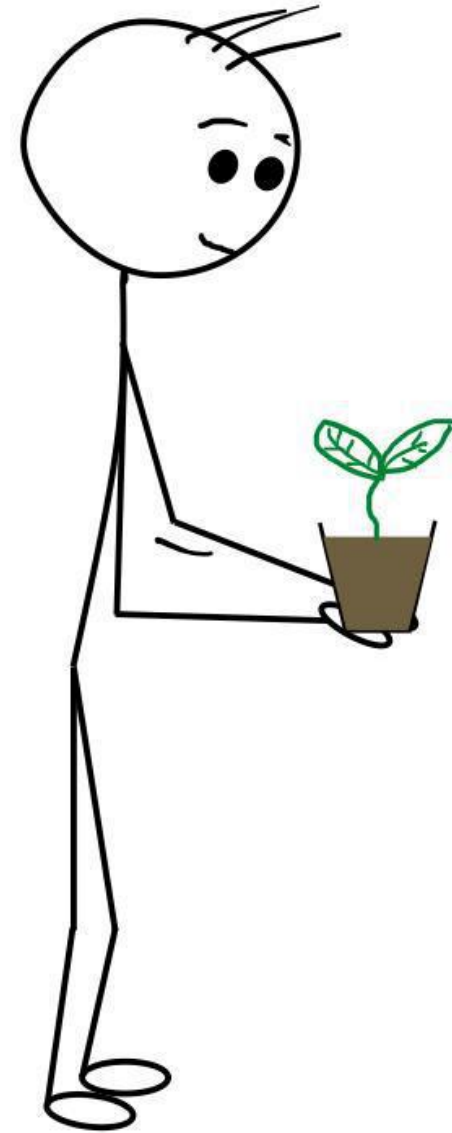
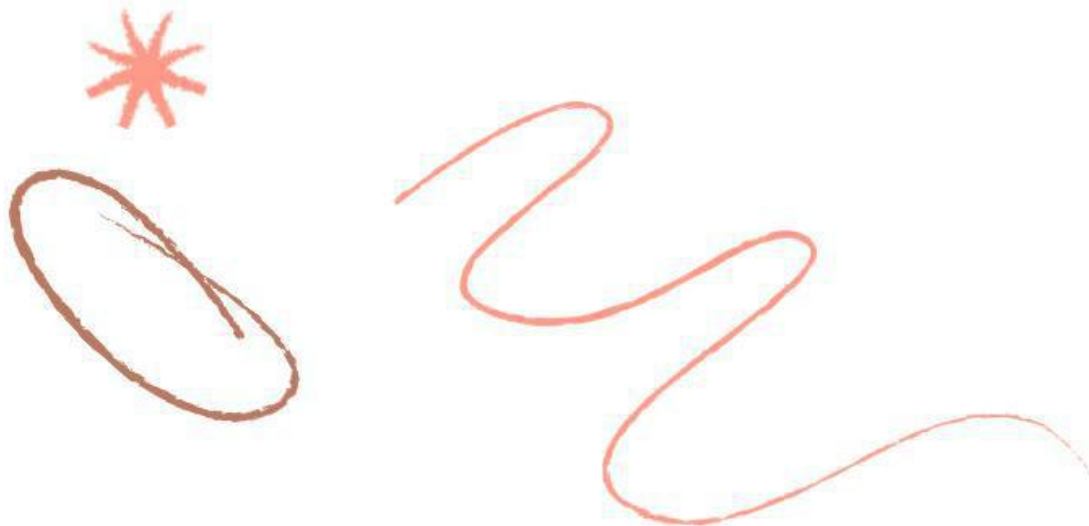


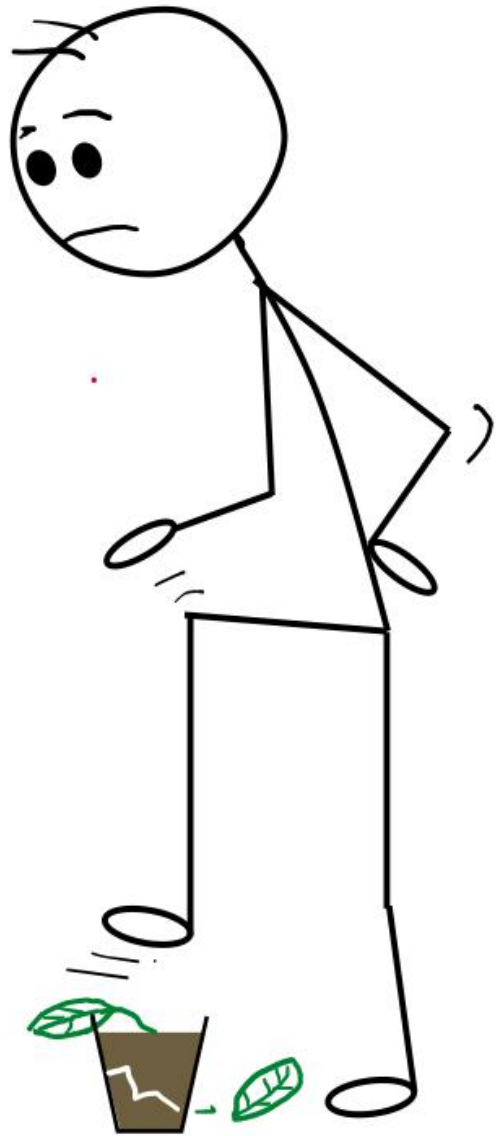
Beliefs and  
behaviours fall into  
two categories





# Growth-Affirming Beliefs & Behaviours

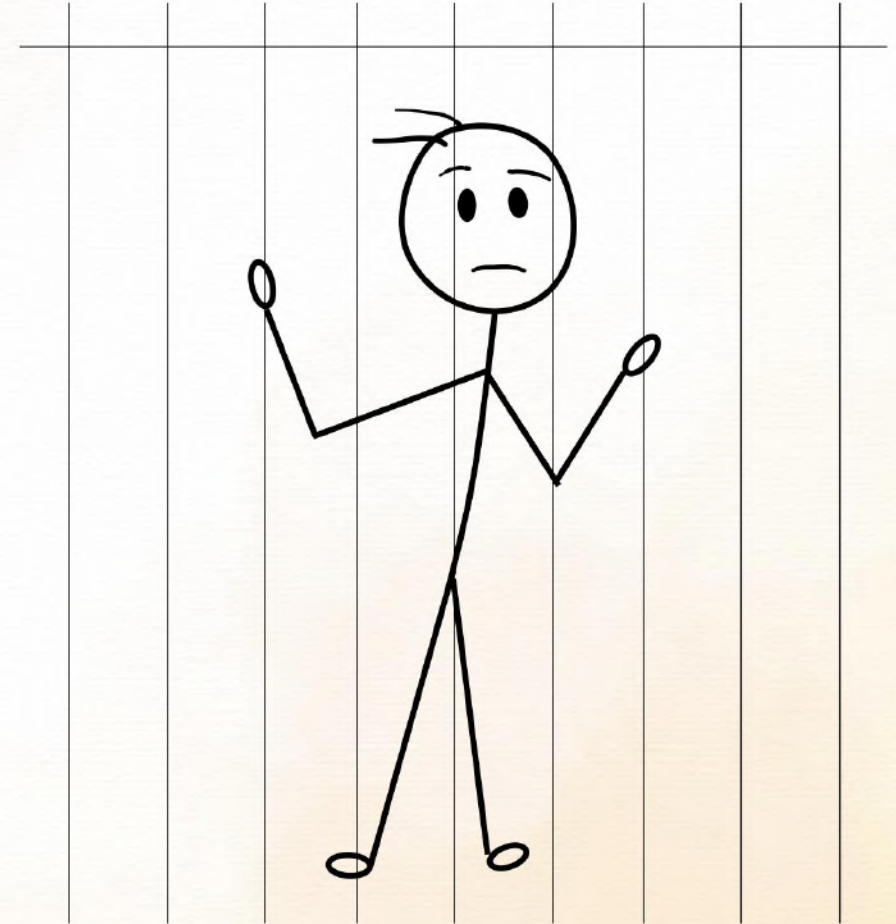




# Growth-Defeating Beliefs & Behaviours



Growth-defeating beliefs  
are a prison cell that locks  
from the inside





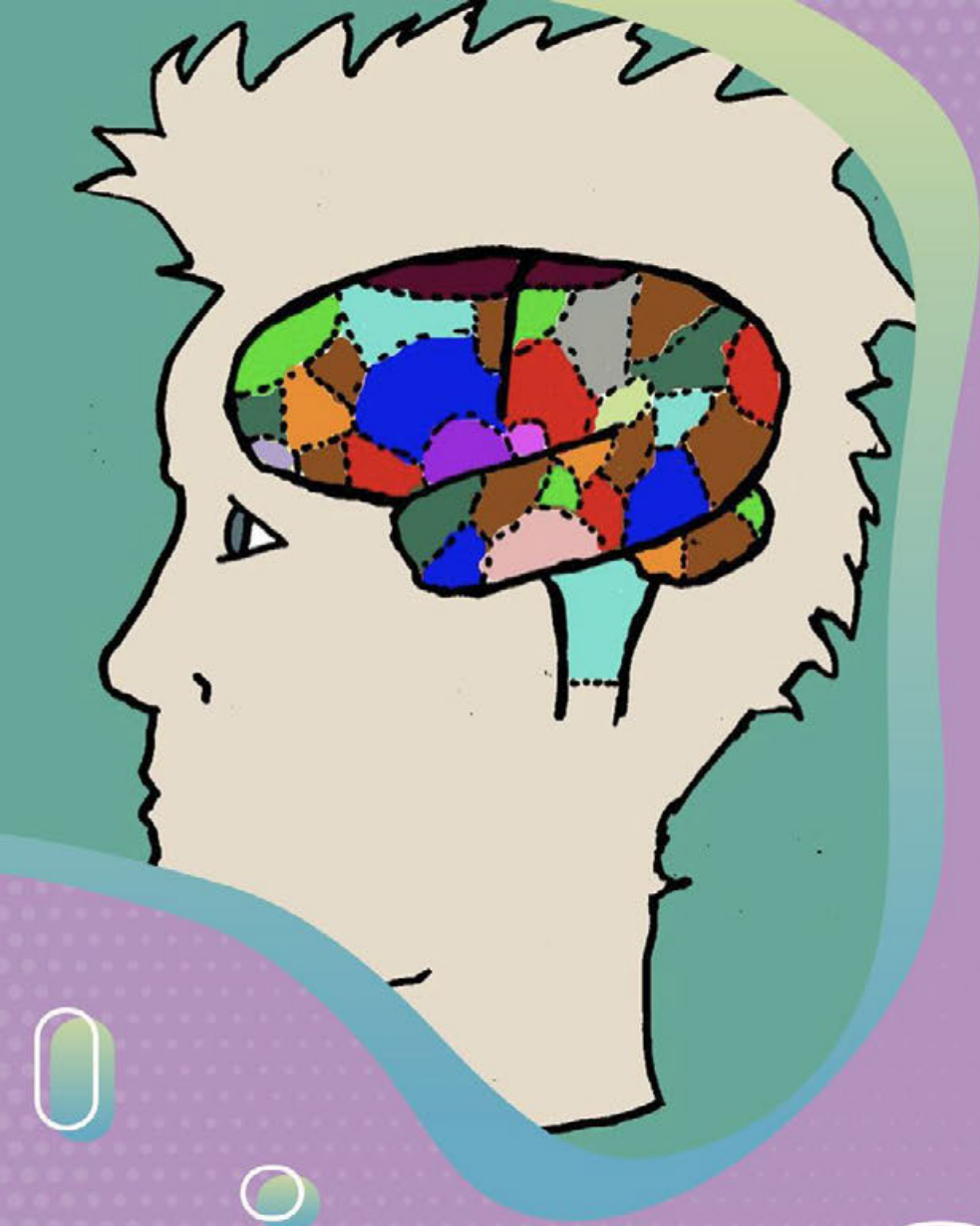
The message from neuroscience  
is clear, we are all brain-  
shapers everyday...





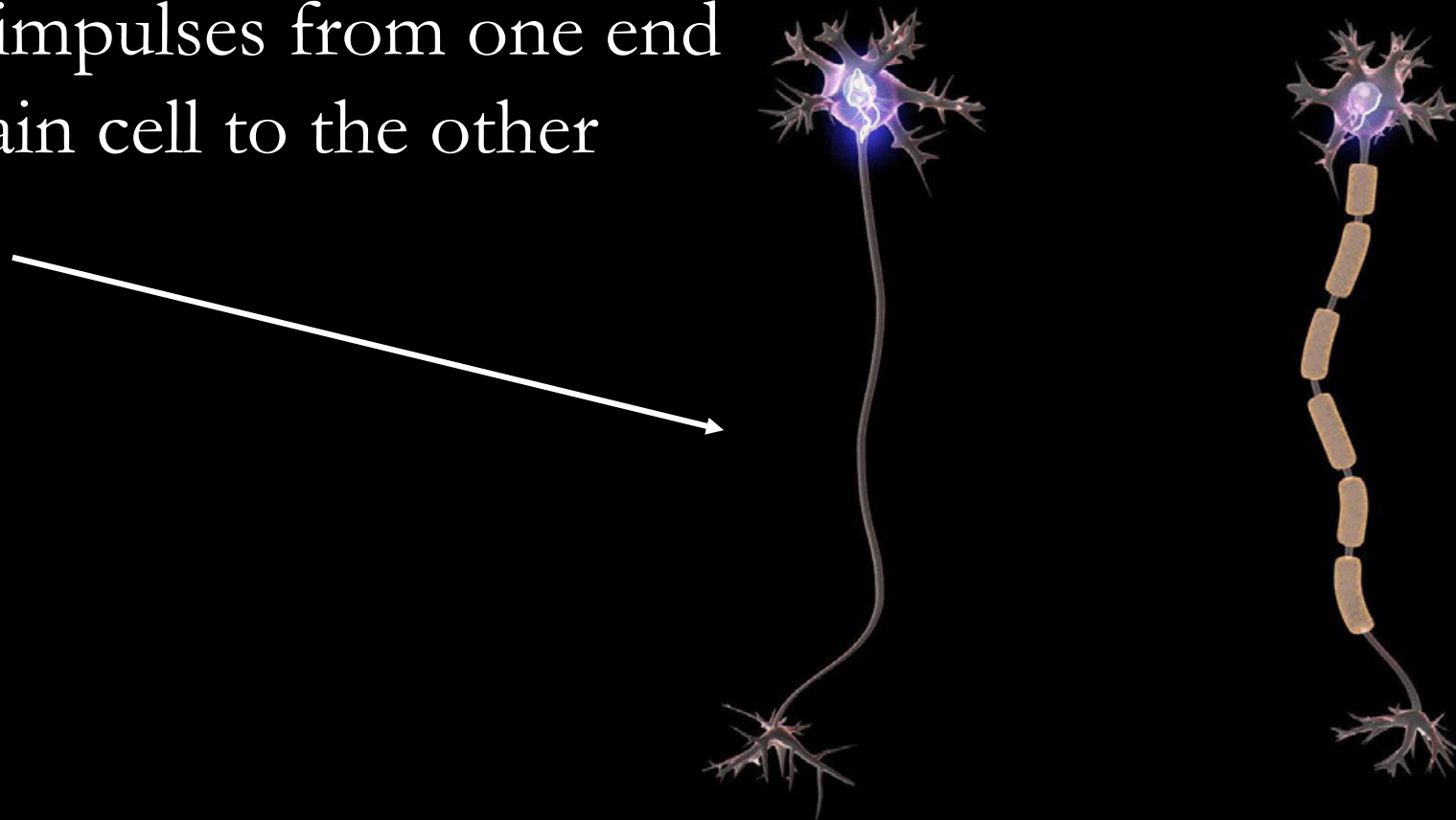
Your brain changes on the basis of  
what you use it for...

If you repeatedly practice something, your brain  
and body make it easier for you to do that thing  
regardless of whether you like that thing, or it is  
good for you...



## Axon

The biological wire that moves electrical impulses from one end of the brain cell to the other





Frazzlement

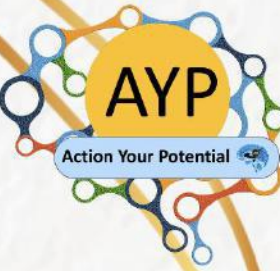


Stress drains blood from the thinking parts of the brain





What are your positive coping strategies? Things we do deliberately to manage the stress and strain we face as our body, brain and mind respond to change in the world.



- Exercising every day
- Protecting your sleep
- Eating healthily
- Sharing your worries rather than bottling them up
- Walking outside
- Listening to music
- Spend time with family
- Making our space tidy

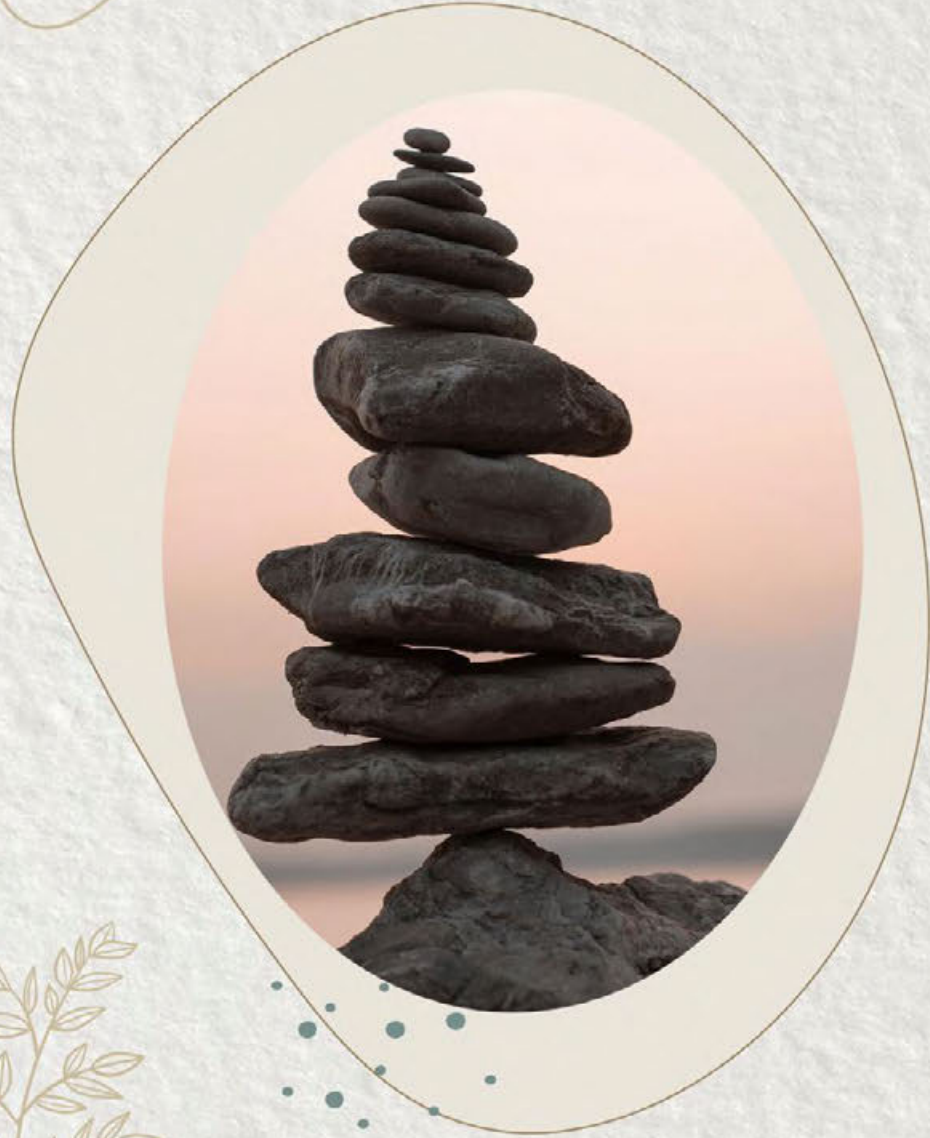
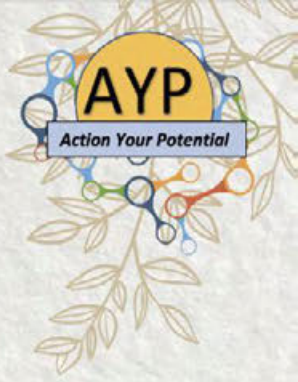


- Playing sport
- Writing a journal
- Being creative
- Practicing our faith
- Social activities
- Dancing
- Learning something new
- Cooking a meal

- Charitable works
- Looking after your pet
- Thinking positively
- Reading a good book
- Breathing exercises
- Mindfulness
- Managing our worries



# The 12 Rocks of Well-Being



Rock 1 - Sleep: 8-9 Hours a night

Rock 2 - Exercise: 20 minutes per day

Rock 3 - Eat and drink healthily

Rock 4 - Mindfulness: 5-10 mins per day

Rock 5 - Mind wandering: allow your mind to problem solve

Rock 6 - Manage emotions: notice, accept, share

Rock 7 - Walk outside in nature

Rock 8 - Listen to music

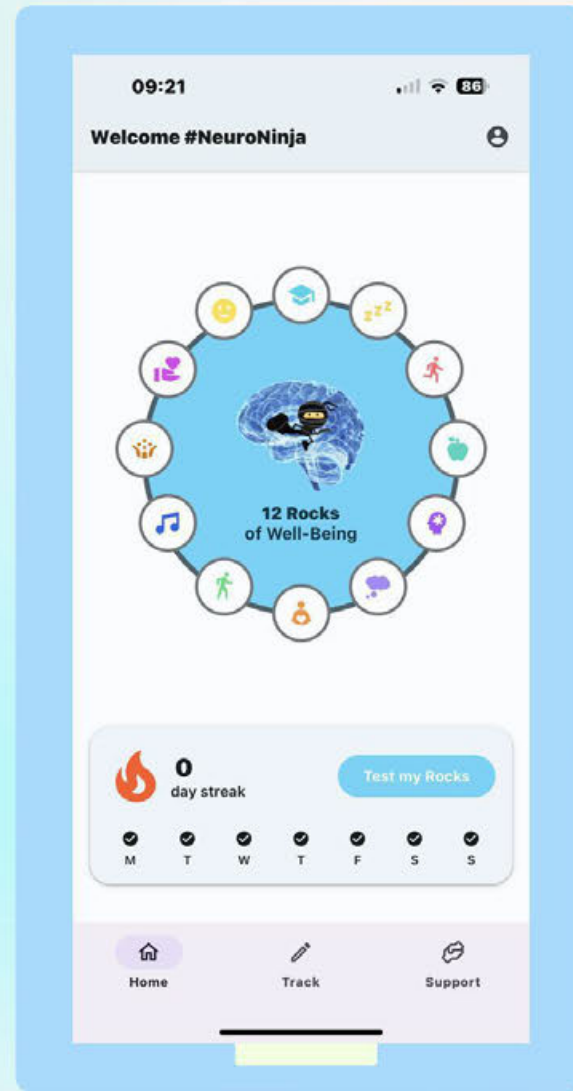
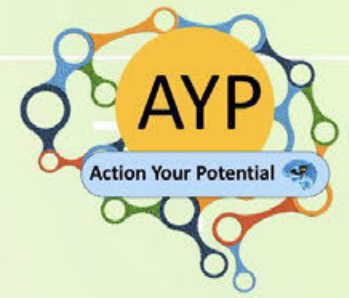
Rock 9 - Connect with friends and family

Rock 10 - Gratitude and Kindness: express both explicitly

Rock 11 - Flow: do what you love

Rock 12 - Learn, play, create, read

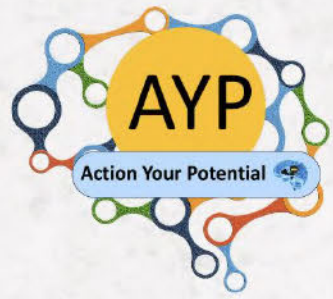
# NeuroNinja App



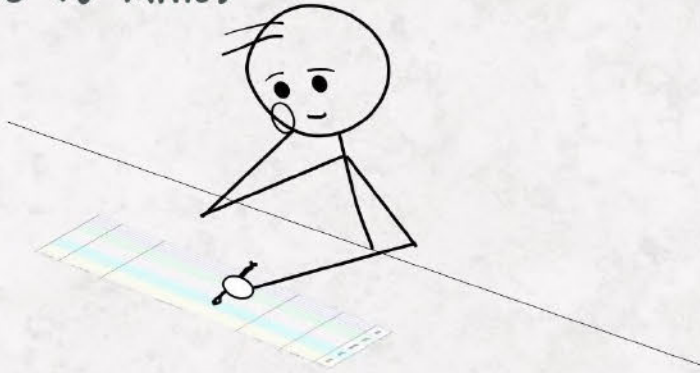
...Available Now...



# KS4 - Learning Routines - Each Day



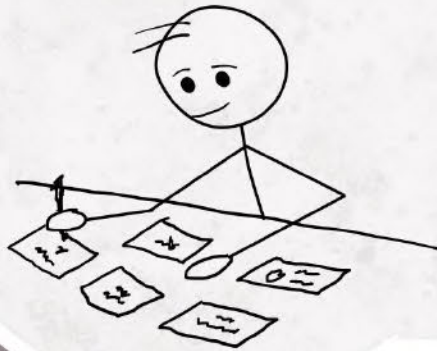
1- Study Capture (5 mins)



3- Mind Map Build/Review (5 mins)



2- Flash Cards (5 mins)

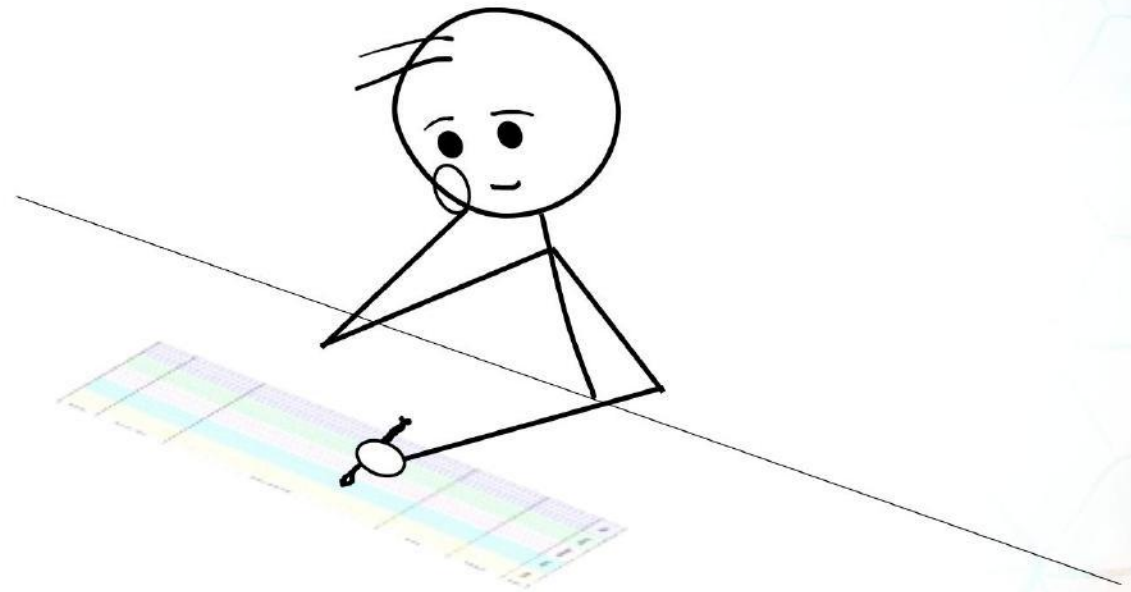


4- Effortful Subjects (10 mins)





# Study Capture



# How we think we Learn...

I've got it - lets  
move on

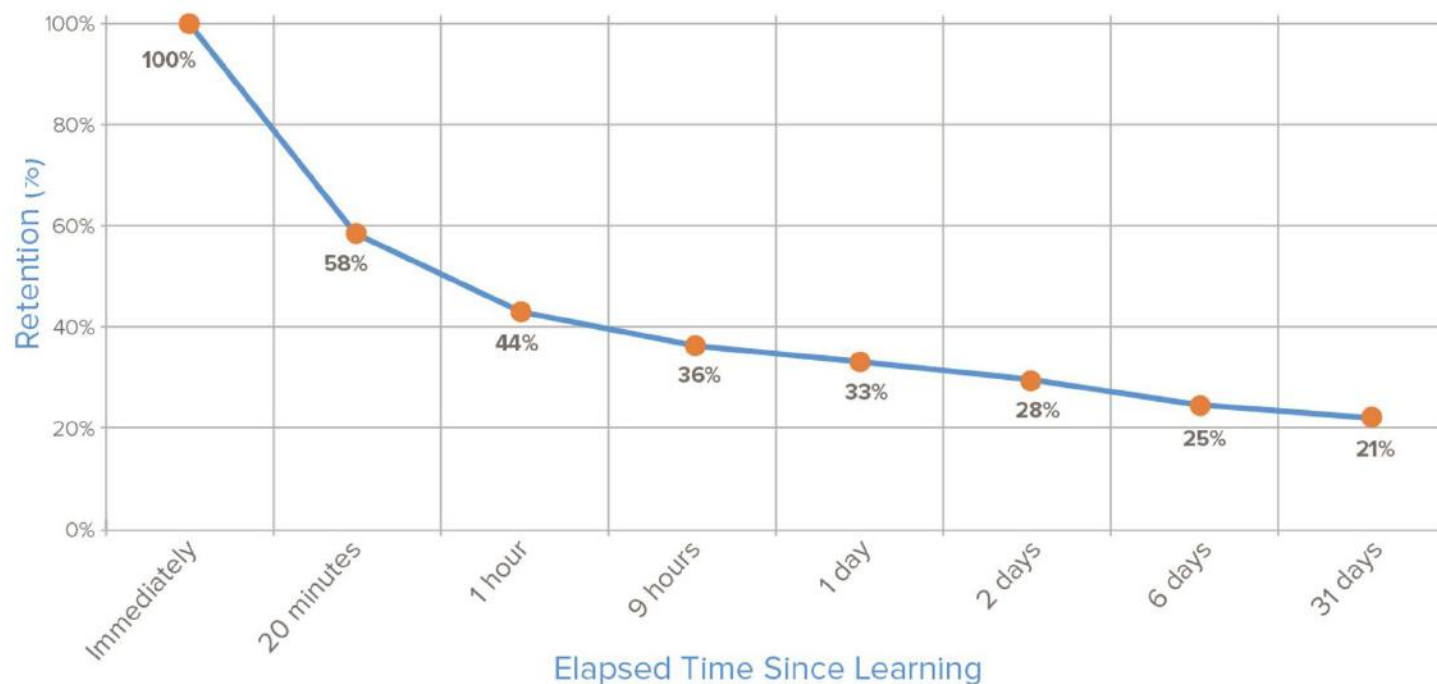
But...

Within 1 Hour - you have forgotten 56% of it

Within 1 day - 67%

Within a Week - 75%

Ebbinghaus Forgetting Curve

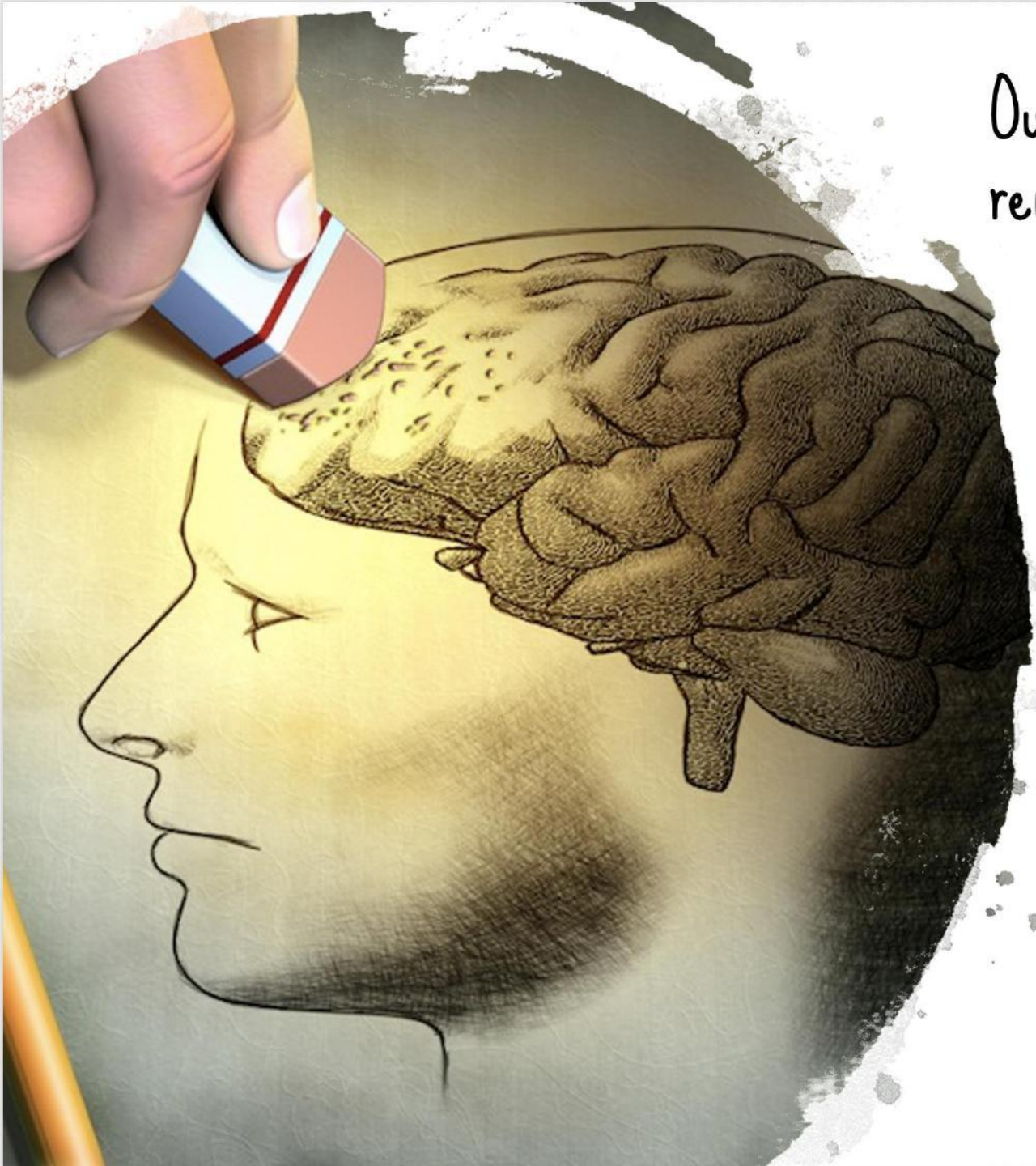




Our Brain finds it easiest to remember information that is;

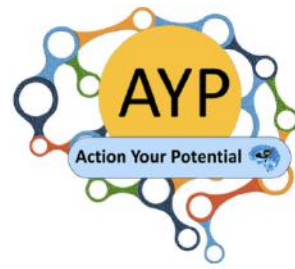
1. Dangerous to us
2. Salient (Interesting to us)
3. Repeated

So...If you don't repeat,  
your brain will delete...



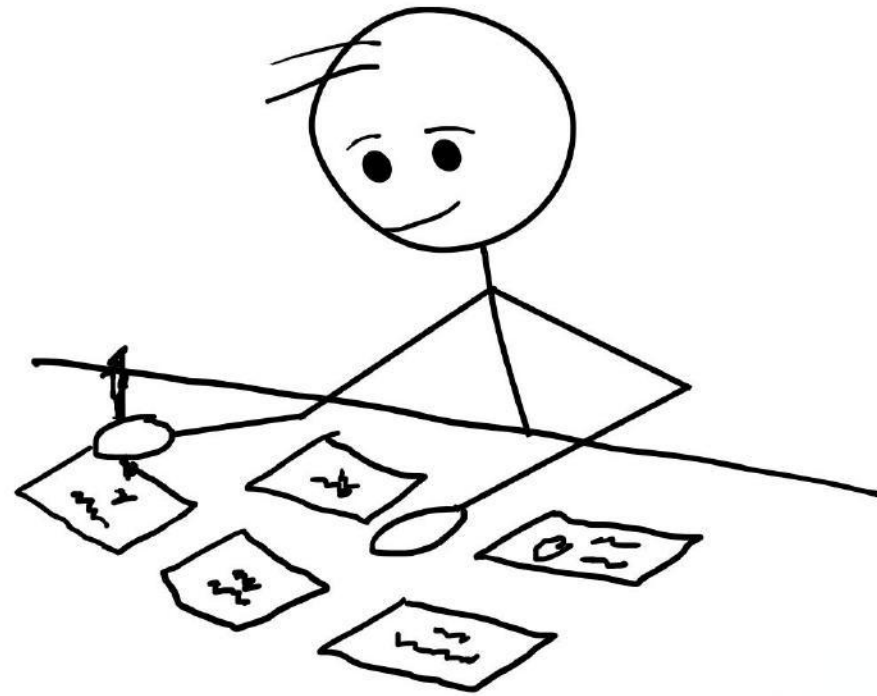


# Study Capture Sheet



Date	Subject	Topic	Key Take away	Key words	Rating
Monday					
Tuesday					
Wednesday					
Thursday					
Friday					

# Flash Cards





# Science Example

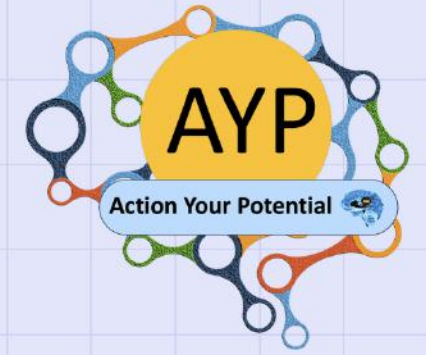
Front

What is  
'current?'

Back

A flow of electrons  
measured in amps  
(A)

# Frayer Flash Cards



Statement

Write a simple statement about what you are making the flashcard about

Elaborate

Give more explanation

Example

Provide any examples

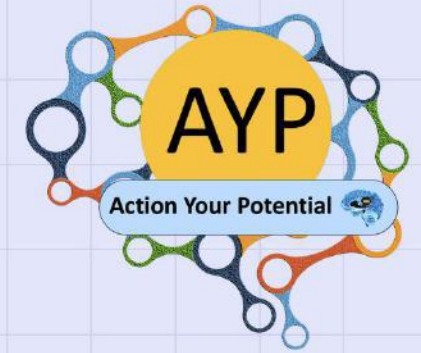
Image

Include a helpful image





# Frayer Flash Cards



EXAMPLE



Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

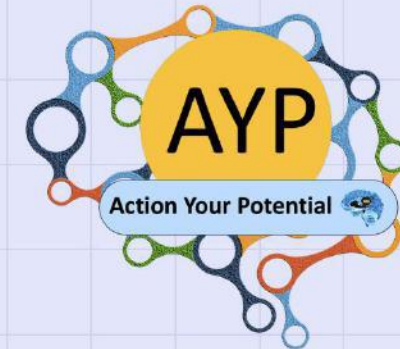
Elaborate

Example

Image



# Frayer Flash Cards



Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

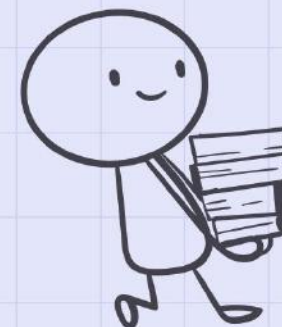
Elaborate

Radioactivity:  
Is emitted from the nuclei of isotopes of unstable atoms. Isotopes are atoms with the same number of protons but different numbers of neutrons in the nucleus. E.g.  $O^{16}$  and  $O^{18}$   
These have unstable nuclei and emit alpha or beta particles, or gamma waves, or neutrons. Alpha particles have 2 protons and 2 neutrons. Are positively charged and are absorbed by paper. Beta particles are electrons from the nucleus. Are negatively charged and absorbed by 2-3cm of aluminium. Gamma radiation is waves of energy emitted from the nucleus. Have no charge and absorbed by thick lead, several cm.

Example

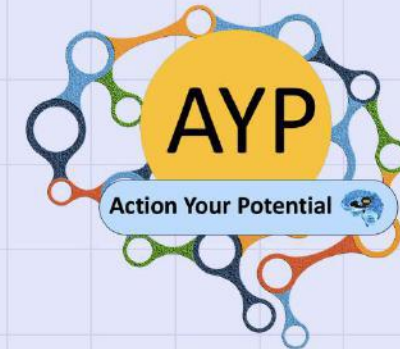
Image

EXAMPLE





# Frayer Flash Cards



Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

Elaborate

Radioactivity:

Is emitted from the nuclei of isotopes of unstable atoms. Isotopes are atoms with the same number of protons but different numbers of neutrons in the nucleus. E.g.  $O^{16}$  and  $O^{18}$

These have unstable nuclei and emit alpha or beta particles, or gamma waves, or neutrons. Alpha particles have 2 protons and 2 neutrons.

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Example

Properties of radiation

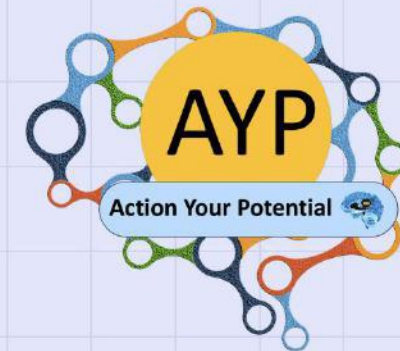
	Alpha	Beta	Gamma
Charge	+2	-1	0
Mass	2	0	0
Ionisation power	High	Middle	Low
Penetration in air	< 5cm	< 1m	< 1km
Stopped by	Paper	Aluminium	Lead

Image

EXAMPLE



# Frayer Flash Cards



EXAMPLE

## Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

## Elaborate

### Radioactivity:

Is emitted from the nuclei of isotopes of unstable atoms. Isotopes are atoms with the same number of protons but different numbers of neutrons in the nucleus. E.g.  $O^{16}$  and  $O^{18}$

These have unstable nuclei and emit alpha or beta particles, or gamma waves, or neutrons. Alpha particles have 2 protons and 2 neutrons.

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## Example

Properties of radiation

	Alpha	Beta	Gamma
Charge	+2	-1	0
Mass	2	0	0
Ionisation power	High	Middle	Low
Penetration in air	< 5cm	< 1m	< 1km
Stopped by	Paper	Aluminium	Lead

## Image

ALPHA PARTICLE



2 PROTONS  
2 NEUTRONS

BETA PARTICLE



ELECTRON

GAMMA RAY



EM WAVE

NEUTRON



NEUTRON

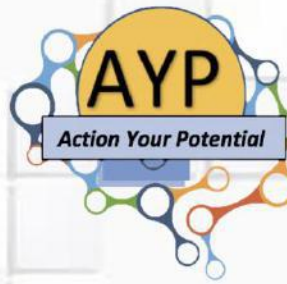




# Flashcards

Day	Colour of flashcard	Subject

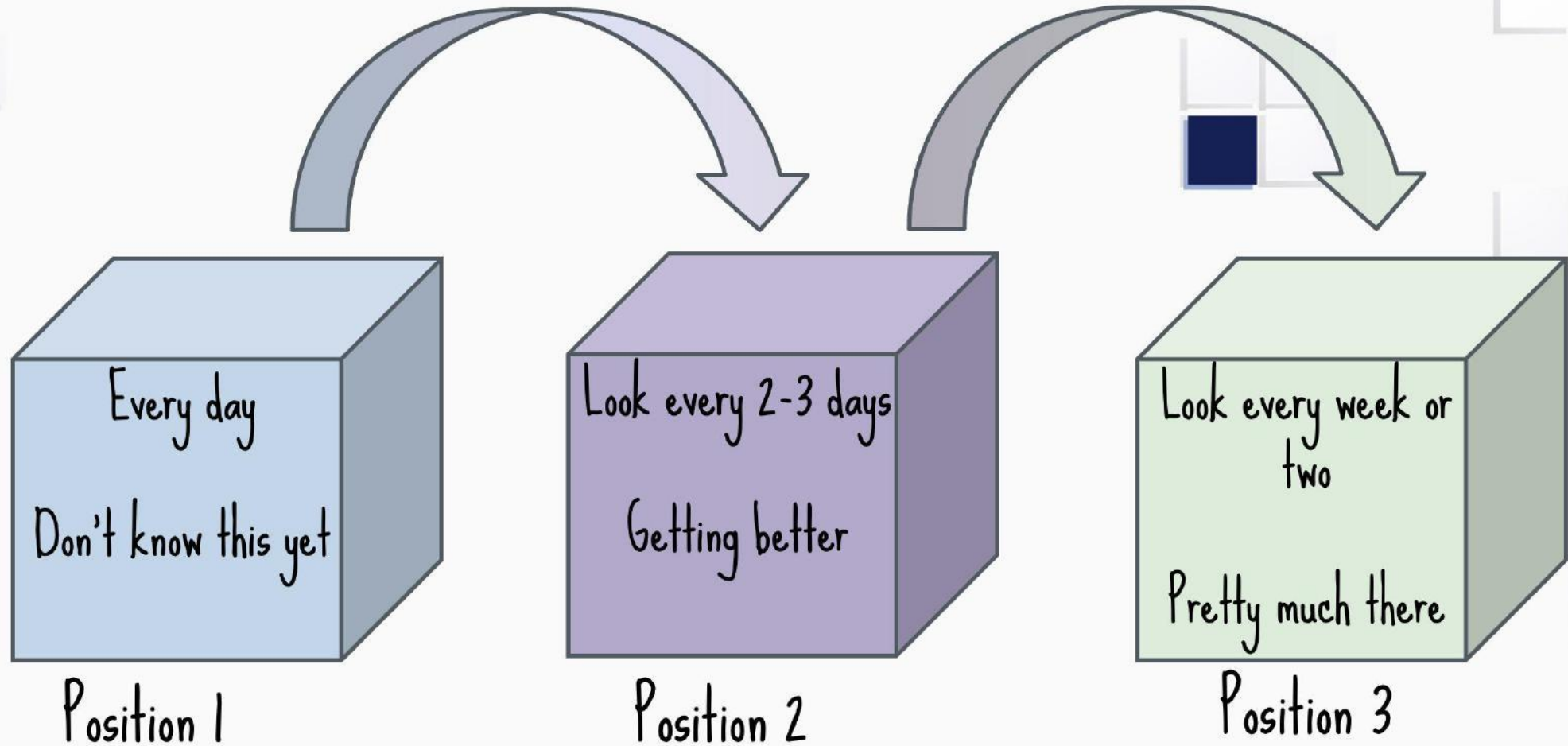
# Flashcards - e.g.



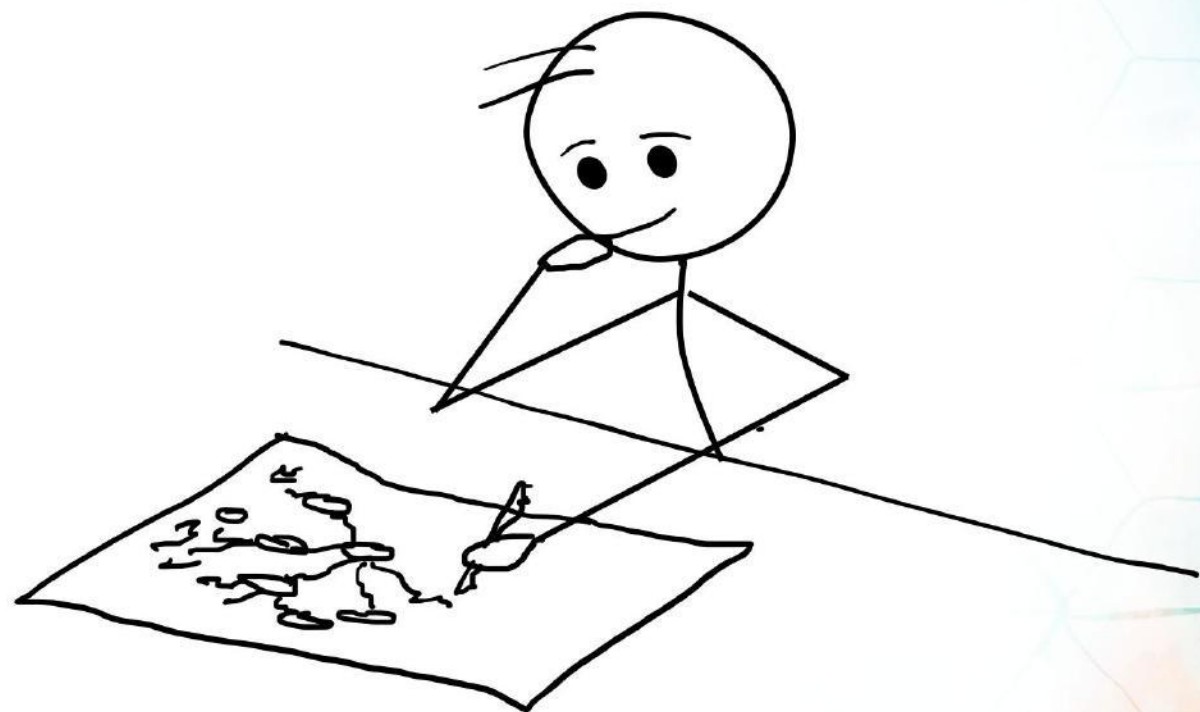
Day	Colour of flashcard	Subject
Monday week 1	Pink	English
Tuesday week 1	Green	Maths
Wednesday week 1	Yellow	French
Thursday week 1	Orange	Geography
Friday week 1	Day off	Day off
Saturday week 1	Blue	Biology
Sunday week 1	Day off	Day off
Monday week 2	Pink	History
Tuesday week 2	Green	PE
Wednesday week 2	Yellow	Art
Thursday week 2	Orange	Chemistry
Friday week 2	Day off	Day off
Saturday week 2	Blue	Physics
Sunday week 2	Day off	Day off



# The box system

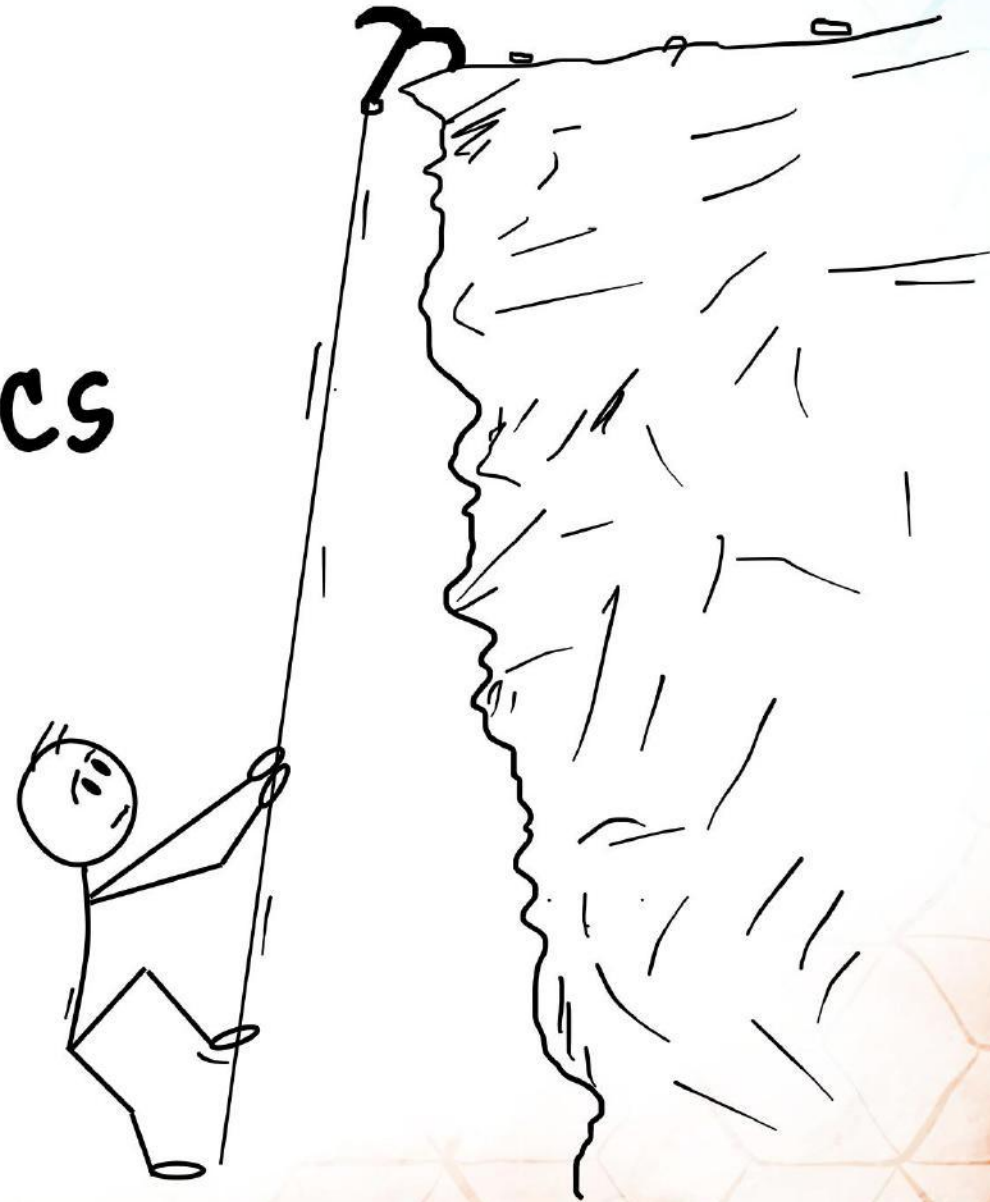


# Mind maps

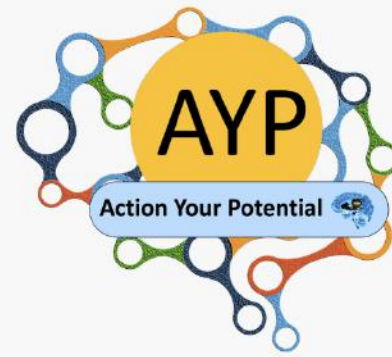




# Effortful Topics



# Study Revolution:

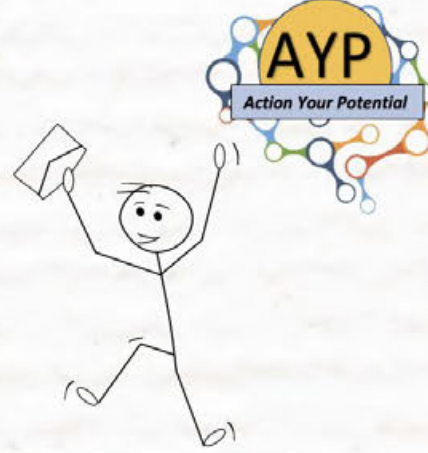


What is the best way to  
revise?





But we build up to that through small steps  
which we're going to work you through.

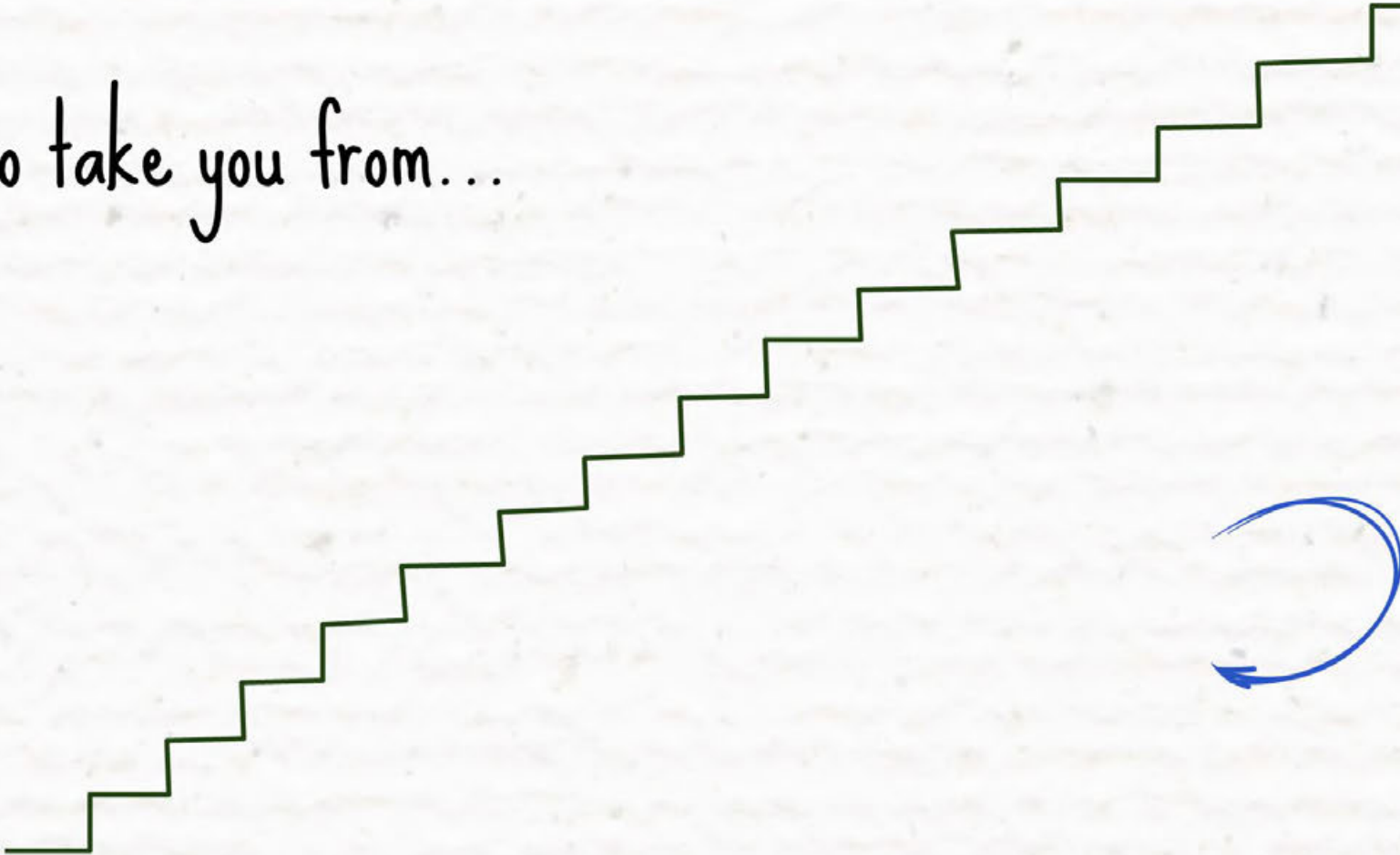


We're going to take you from...

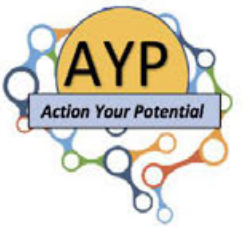
to Study Hero...



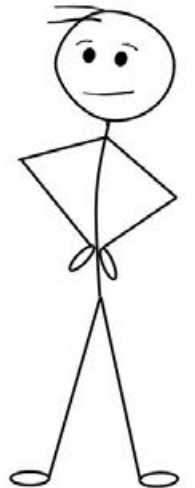
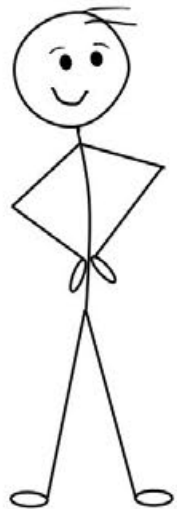
Study Zero



# Confidence Chart -GCSE

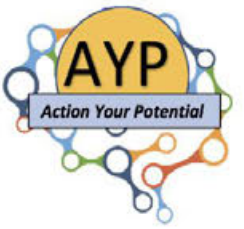


Subject	Grade You Want	Grade Stretch Goal	Current Confidence level		
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️
			😊	😐	☹️

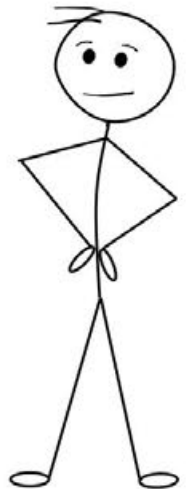
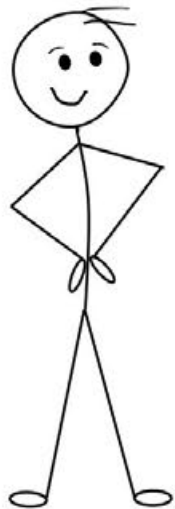




# Confidence Chart - GCSE



Subject	Grade You Want	Grade Stretch Goal	Current Confidence level
Maths	5	6	☺ ☹ ☺
English Lang	5	6	☺ ☹ ☹
English Lit	5	6	☺ ☹ ☹
Chemistry	6	7	☺ ☹ ☹
Physics	5	6	☺ ☹ ☹
Biology	6	7	☺ ☹ ☹
French	4	5	☺ ☹ ☹
Geography	6	7	☺ ☹ ☹
Tech	6	7	☺ ☹ ☹
RE	4	5	☺ ☹ ☹
			☺ ☹ ☹
			☺ ☹ ☹



# What is a study rep?

3 types of rep



Learning rep

Do this when you don't understand  
or don't remember a topic

Practice rep

Do this when it's a bit hazy

Testing rep

Do this when you need to test  
yourself

# What is it all about?

**Repetition** - a carefully designed **study slot** that works with the memory and learning systems of our brains to achieve **maximum impact** for learning, understanding and recall

## Key Attitudes

**Grit** - Keeping going

**Attention** - 1 thing at a time

**Focus** - don't get distracted internally or externally

**Attitude** - Can do

**Mode** - taking responsibility

How long does it take?

≡ **25 minutes** ≡






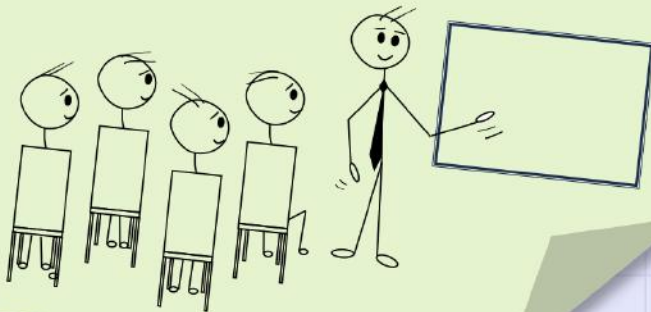
# How does learning work at the level of the brain?



Learning is 3  
processes



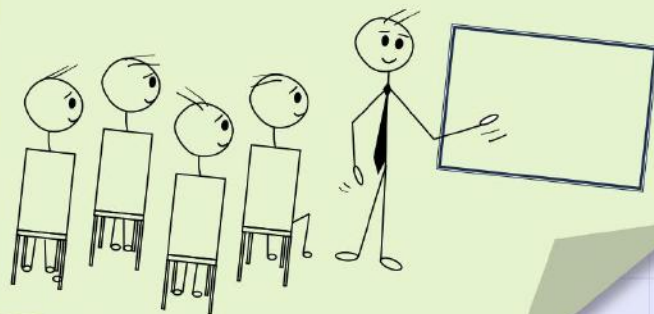
Getting it - understanding  
what you have been taught



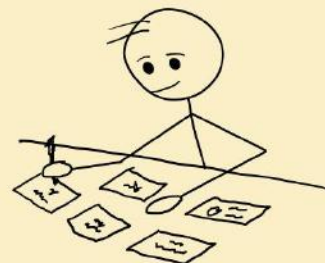
# How does learning work at the level of the brain?

Learning is 3 processes

**Getting it** - understanding what you have been taught



**Practicing it** - Encoding it in the brain

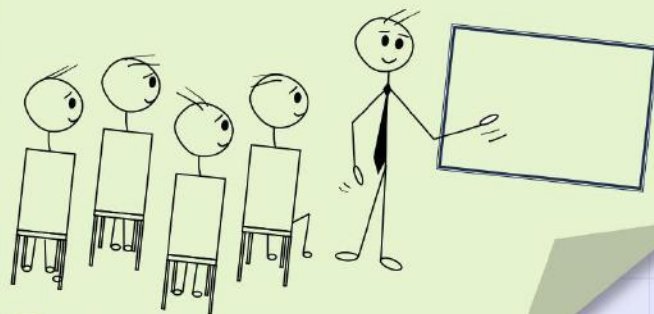




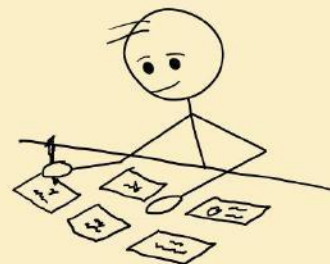
# How does learning work at the level of the brain?

Learning is 3  
processes

**Getting it** - understanding  
what you have been taught



**Practicing it** - Encoding it  
in the brain

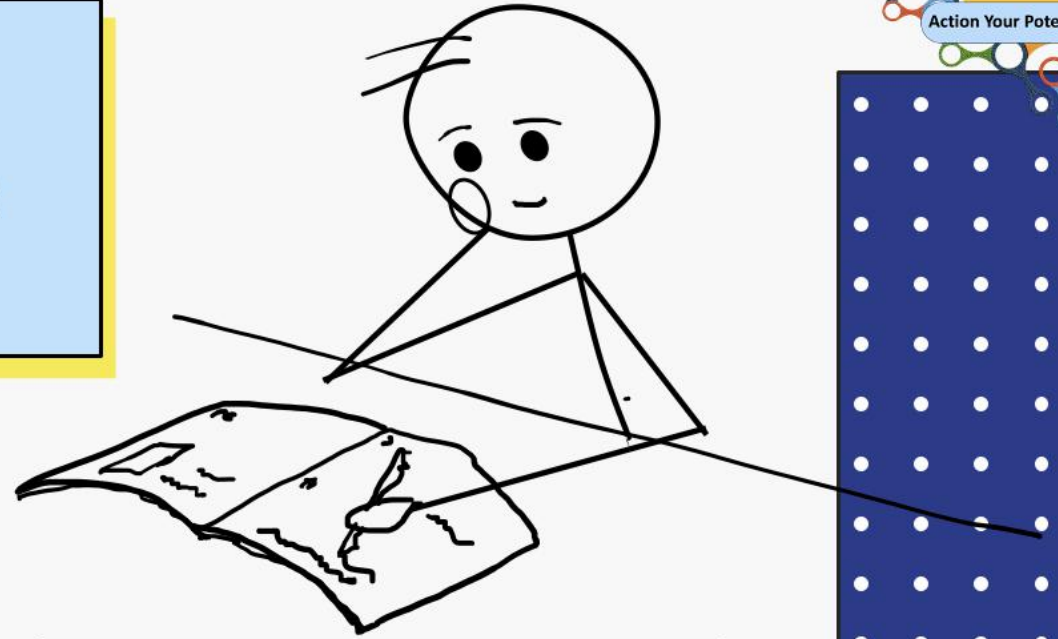


**Using it** - Applying the  
new learning





# Study Revolution:



Taking Responsibility for what  
you know using study reps

example

1 Choose a topic and write  
down what you know about it

Topic - Radioactivity

5 mins

- Something to do with atoms
- Small particles
- Waves?
- Can harm you
- Alpha
- Nuclear

Learning Rep

example

1 Choose a topic and write down what you know about it

Topic - Radioactivity

5 mins

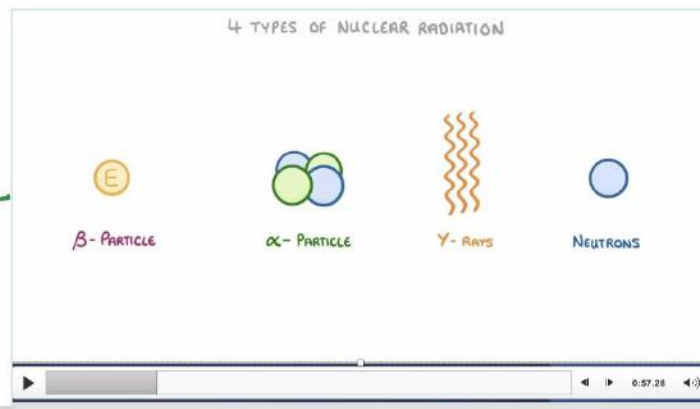
- Something to do with atoms
- Small particles
- Waves?
- Can harm you
- Alpha
- Nuclear

# Learning Rep

2

10 mins

Study the resource





example

1 Choose a topic and write down what you know about it

Topic - Radioactivity

5 mins

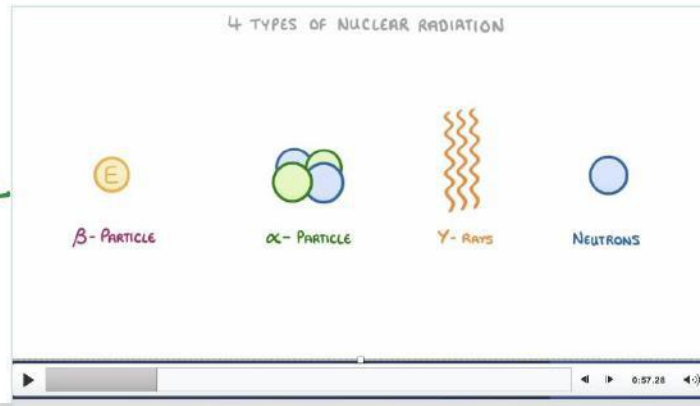
- Something to do with atoms
- Small particles
- Waves?
- Can harm you
- Alpha
- Nuclear

# Learning Rep

2

10 mins

Study the resource



3

"Summarise by taking notes whilst using the resource... Video, Revision Guide, Class Notes.."

Radioactivity: Emitted from unstable atoms called isotopes.  
Isotopes = Atoms with the same number of protons, but different numbers of neutrons

Alpha particles: 2 protons and 2 neutrons. Quite large particles. Travel a small distance. Absorbed by a thin layer of paper

Beta particles: A small negatively charged particle from the nucleus. A neutron splits into a proton and an electron. Have less mass and travel further in air. Absorbed by a thin layer of Aluminium

Gamma waves (rays): A wave of energy released from the nucleus of an unstable isotope. Absorbed by very thick lead or metres of concrete

Neutrons: Emitted by an unstable isotope with too many neutrons in the nucleus. Can cause a chain reaction.

10 mins

example

1 Choose a topic and write  
down what you know about it

Topic - Radioactivity 5 mins

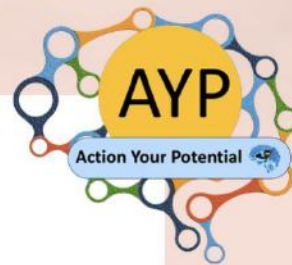
What are the key ideas?

- Alpha: 2 neutrons and 2 protons
- Beta: negative electron form the nucleus
- Gamma: Wave of energy
- Alpha least penetrating, gamma most
- Alpha most ionising, gamma least ionising
- Neutrons are released from unstable nuclei

Practice Rep



example



1 Choose a topic and write down what you know about it

Topic - Radioactivity 5 mins

What are the key ideas?

- Alpha: 2 neutrons and 2 protons
- Beta: negative electron form the nucleus
- Gamma: Wave of energy
- Alpha least penetrating, gamma most
- Alpha most ionising, gamma least ionising
- Neutrons are released from unstable nuclei

2

Make flashcards, mind map or summary notes.

Radioactivity: Particles and waves emitted from the nuclei of isotopes of unstable atoms.

15 mins

## Practice Rep

Isotope: Atoms with the same number of protons but different numbers of neutrons in the nucleus e.g.,  $O^{16}$  and  $O^{18}$

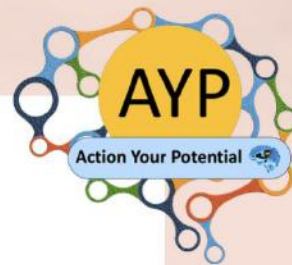
Alpha radiation: Particles that have 2 protons and 2 neutrons. Are positively charged. Absorbed by paper.

Beta radiation: Are electrons from the nucleus. Are negatively charged. Absorbed by 2-3cm of aluminium.

Gamma radiation: Waves of energy emitted from the nucleus. Have no charge. Absorbed by thick lead, several cm.



example



1 Choose a topic and write down what you know about it

Topic - Radioactivity 5 mins

What are the key ideas?

- Alpha: 2 neutrons and 2 protons
- Beta: negative electron form the nucleus
- Gamma: Wave of energy
- Alpha least penetrating, gamma most
- Alpha most ionising, gamma least ionising
- Neutrons are released from unstable nuclei

3 Read through  
5 mins

2 Make flashcards, mind map or summary notes.

Radioactivity: Particles and waves emitted from the nuclei of isotopes of unstable atoms.

15 mins

## Practice Rep

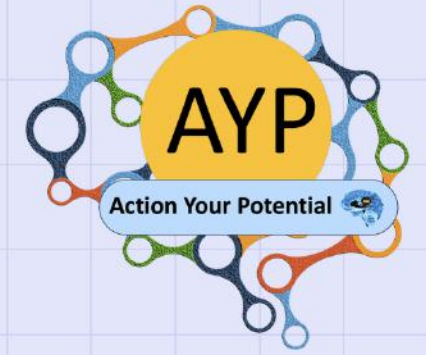
Isotope: Atoms with the same number of protons but different numbers of neutrons in the nucleus e.g.,  $O^{16}$  and  $O^{18}$

Alpha radiation: Particles that have 2 protons and 2 neutrons. Are positively charged. Absorbed by paper.

Beta radiation: Are electrons from the nucleus. Are negatively charged. Absorbed by 2-3cm of aluminium.

Gamma radiation: Waves of energy emitted from the nucleus. Have no charge. Absorbed by thick lead, several cm.

# Frayer Flash Cards



Statement

Write a simple statement about what you are making the flashcard about

Elaborate

Give more explanation

Example

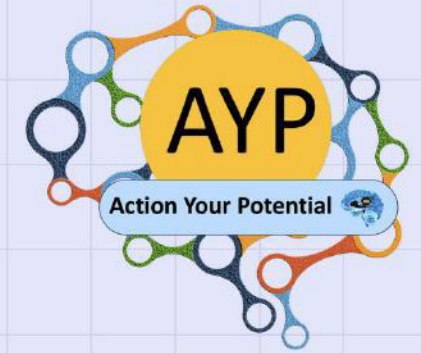
Provide any examples

Image

Include a helpful image



# Frayer Flash Cards



EXAMPLE



Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

Elaborate

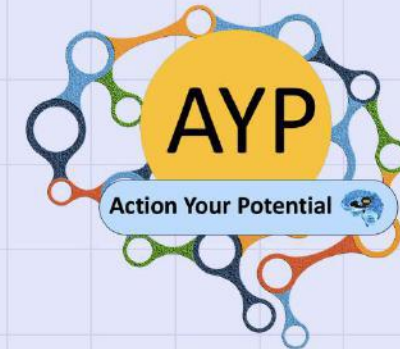
Example

Image





# Frayer Flash Cards



EXAMPLE



Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

Elaborate

Radioactivity:

Is emitted from the nuclei of isotopes of unstable atoms. Isotopes are atoms with the same number of protons but different numbers of neutrons in the nucleus. E.g.  $O^{16}$  and  $O^{18}$

These have unstable nuclei and emit alpha or beta particles, or gamma waves, or neutrons. Alpha particles have 2 protons and 2 neutrons.

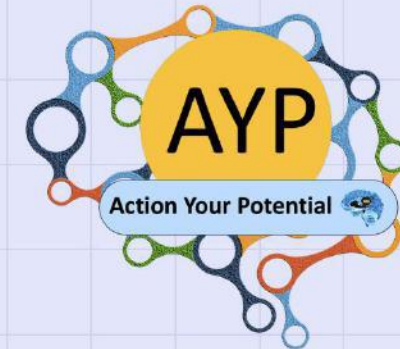
Are positively charged and are absorbed by paper. Beta particles are electrons from the nucleus. Are negatively charged and absorbed by 2-3cm of aluminium. Gamma radiation is waves of energy emitted from the nucleus. Have no charge and absorbed by thick lead, several cm.

Example

Image



# Frayer Flash Cards



Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

Elaborate

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Example

Properties of radiation

	Alpha	Beta	Gamma
Charge	+2	-1	0
Mass	2	0	0
Ionisation power	High	Middle	Low
Penetration in air	< 5cm	< 1m	< 1km
Stopped by	Paper	Aluminium	Lead

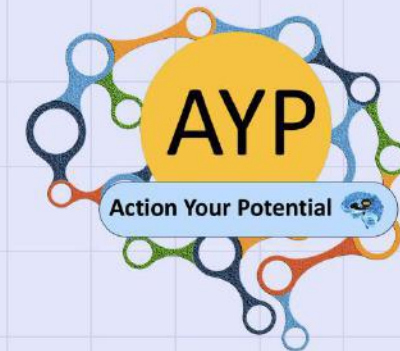
Image

EXAMPLE





# Frayer Flash Cards



EXAMPLE

## Statement

Radioactivity:  
Particles and waves emitted from  
the nuclei of isotopes of unstable  
atoms.

## Elaborate

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## Image

ALPHA PARTICLE



2 PROTONS  
2 NEUTRONS

BETA PARTICLE



ELECTRON

GAMMA RAY



EM WAVE

NEUTRON



NEUTRON





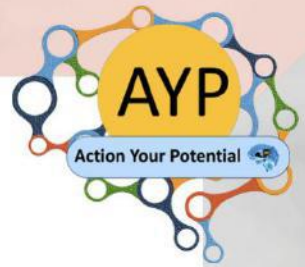
example

1 Find the past paper  
question and mark scheme

Topic - Radioactivity

5 mins

Testing Rep



example

1

Find the past paper question and mark scheme

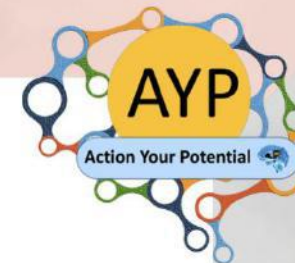
Topic - Radioactivity

5 mins

Testing Rep

2

Have a go at the question



The chart below shows the change in mass number and atomic number of an atom for different types of radioactive decay.

	change in atomic number	change in mass number
A	-1	0
B	-2	-4
C	0	0
D	+1	0

(a) Write down the letter which shows how the mass number and atomic number change for the following:

(i) emission of an alpha particle  
.....[1]

(ii) emission of a beta particle  
.....[1]

(iii) emission of gamma radiation  
.....[1]

(b) Complete the equation below to show the decay of  $^{238}_{92}\text{U}$  to an isotope of thorium (Th) when it emits an alpha particle. [2]



(c)  $^{238}_{92}\text{U}$  and  $^{235}_{92}\text{U}$  are both forms of uranium.

(i) What do we call different forms of the same element?  
.....[1]

(ii) Describe how their nuclei differ from each other.  
.....[1]

15 mins

example

1 Find the past paper question and mark scheme

Topic - Radioactivity

5 mins

# Testing Rep

3 Use the mark scheme to add what was missed.

5 mins

2

Have a go at the question

The chart below shows the change in mass number and atomic number of an atom for different types of radioactive decay.

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.....[1]

15 mins



# What is a study rep?

3 types of rep

Learning rep

Do this when you don't understand or don't remember a topic

Practice rep

Do this when it's a bit hazy

Testing rep

Do this when you need to test yourself

# Your Learning Day Plan

We need to balance study, well-being and mind management



The plan should have these elements

Learning and practice reps for each subject- (green)

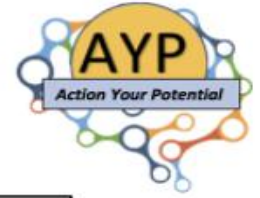
Testing reps for each subject- (orange)

Well-being time- (yellow)

Personal time- (purple)

Planning slot- (grey)

# General Planner - Week 1



Time	Monday	Tuesday	Wednesday	Thursday	Friday	Saturday	Sunday
7-8am							
8-9am							
9-10am							
10-11am							
11-12pm							
12-1pm							
1-2pm							
2-3pm							
3-4pm							
4-5pm							
5-6pm							
6-7pm							
7-8pm							
8-9pm							
9-10pm							
10-11pm							







Be in no doubt you  
can do this...